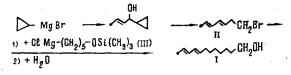
INSECT PHEROMONES AND THEIR ANALOGS

I. SYNTHESIS OF THE SEX ATTRACTANT OF THE CODLING MOTH

```
G. A. Tolstikov, U. M. Dzhemilev,
and R. I. Khusnutdinov
```

The study of copolymerization reactions of 1,3-dienes with acrylates catalyzed by transition metals is opening up new possibilities for the stereochemical synthesis of unsaturated compounds close in structure to insect sex attractants and juvenoids [1-5]. In the present paper we describe a simple method for the synthesis of the sex attractant of the codling moth Laspeyresia pomonella.

According to Descoins and Henrick [6], the synthesis of the attractant, which has the structure of dodecatrans-8,trans-10-dien-1-ol (I) is formed by a "7+5" scheme. Difficultly accessible cyclopropane derivatives are used to obtain the C₇ fragment. The concluding stage of the synthesis consists of the coupling of 1-bromohepta-3,5-diene (II) with the organomagnesium reagent (III), which is readily obtainable from tetrahydropyran. The whole scheme of synthesis appears as follows:



Scheme 1

We have developed a new scheme for the synthesis of the attractant which provides for the synthesis of the C_7 fragment from accessible compounds. As is well known [7, 3], the copolymerization of butadiene with methyl acrylate catalyzed by iron complexes permits the production of mixture of hepta-cis-2,cis-5-dienoic and hepta-cis-2,trans-5-dienoic acids (IV and V). Their isomerization with methanolic alkali or sodium methanolate gave methyl hepta-trans-3,trans-5-dienoate (VI), which was smoothly reduced by LiAlH₄ to hepta-trans-3,trans-5-dien-1-ol (VIII). The latter, by treatment with PBr₃ was converted into the bromide (II). To pass to the attractant, we used the condensation of the bromide with compound (III) described by Descoins and Henrick [6]. Thus, the method for the synthesis that we propose is based on the use of accessible initial substances and ensures a fairly high yield of attractant (not less than 60% calculated on the mixture of esters (IV) and (V)):

 $\mathcal{H} + \mathcal{A}_{CO_{2}}CH_{3} \xrightarrow{Fe^{0}-L} \mathcal{C}_{CO_{2}}CH_{3} + \mathcal{A}_{V}CO_{2}CH_{3}$ $(IV+V) \xrightarrow{OH^{-}} \mathcal{C}_{O_{2}}CH_{3} \xrightarrow{LiAe H_{4}} \mathcal{A}_{VII} \xrightarrow{VI} CH_{2}OH \xrightarrow{P Br_{3}} \mathcal{H}_{I}$ $(IV+V) \xrightarrow{(II)} \frac{11+(III)}{21+H_{2}O} \qquad (I)$

Scheme 2

EXPERIMENTAL

The methyl esters of hepta-cis-2,cis-5-, -trans-2,cis-5-, and -trans-3,-trans-5-dienoic acids (IV, V, and VI, respectively), and also hepta-trans-3,-trans-5-dieniol (VII) were obtained as described previously [7]. The PMR spectra were recorded on a Tesla BS-480 B instrument (internal standard hexamethyldisiloxane, solvent CCl₄). The IR spectra were recorded on a UR-20 spectrophotometer (film).

 $\frac{1-\text{Bromohepta-trans-3,trans-5-diene (II).}}{\text{of benzene was treated with 3-4 drops of }48\% \text{ HBr and, with stirring at 10°C, 5.5 g of PBr₃ was added}$

Institute of Chemistry, Bashkir Branch of the Academy of Sciences of the USSR, Ufa. Translated from Khimiya Prirodnykh Soedinenii, No. 1, pp. 125-127, January-February, 1978. Original article submitted September 16, 1977.

0009-3130/78/1401-0101\$07.50©1978 Plenum Publishing Corporation

101

UDC 547.364

over 1 h. The reaction mixture was left overnight at room temperature and was then washed with 10% NaHCO₃ solution and with water. The organic layer was dried with Na₂SO₄, the benzene was evaporated off, and the residue was distilled under a reduced pressure of argon. This gave 5.6 g of (II) (yields ~ 80%) with bp 34°C/ 1 mm, n_D²⁰ 1.5200; UV spectrum: $\lambda _{max}^{ethanol}$ 321 nm, ε 1800; IR spectrum (ν , cm⁻¹): 1630, 1665 (C=C-C=C), 975 (trans-CH=CH); PMR spectrum (δ , pp m): 1.68 (CH₃-C=), 5.2-6.3 (CH=CH-CH=CH), 3.20 (-CH₂-Br); m/e 175.

Dodeca-trans-8,trans-10-dienol (1). Over two hours at -5 to 0°C, 4.4 g of 1-bromohepta-trans-3,trans-5-diene (II) was added to a solution of the Grignard reagent $CIMg - (CH_2)_5 - OSi(CH_3)_3$ (III) in tetrahydrofuran (prepared from 0.61 g of Mg and 5 g of $Cl - (CH_2)_5 - OSi(CH_3)_3$ by a known method [6]) and catalytic amounts of the complex Li₂[CuCl₄] [8]. Then the reaction mixture was left at room temperature for 14-15 h, the precipitate that had deposited was filtered off, the tetrahydrofuran was evaporated off, and, without purification, the residue was boiled in aqueous ethanol for 2 h. After cooling, the product was extracted with ether, the extract was dried with Na₂SO₄, the ether was evaporated off, and the residue was distilled in vacuum. This gave 5.4 g of (I), yield 72%. bp 122°C/0.5 mm, mp 29-30°C (recrystallization from pentane). UV spectrum: $\lambda \underset{max}{\text{ethanol}}$ 234 nm; IR spectrum (ν , cm⁻¹): 990, 1060, 3380; PMR spectrum (δ , ppm): 1.65 (CH₃-C=), 2.05 (-CH₂-C=), 3.63 (-CH₂-O-), 5.2-6.30 (CH = CH - CH = CH); m/e 182.

SUMMARY

The synthesis of the sex attractant of the codling moth <u>Laspeyresia pomonella</u> has been performed using as the initial compounds linear dimers of butadiene with methyl acrylate.

LITERATURE CITED

- 1. G. A. Tolstikov, U. M. Dzhemilev, and R. I. Khusnutdinov, Izv. Akad. Nauk SSSR, Ser. Khim., 1562 (1975).
- 2. G. A. Tolstikov, U. M. Dzhemilev, R. I. Khusnutdinov, and L. M. Zelenova, Zh. Org. Khim., <u>12</u>, No. 2, 315 (1976).
- 3. H. Singer, U. Wilfried, and D. Manfred, Synthesis, 5, 265 (1971).
- 4. O. S. Vostrikova, U. M. Dzhemilev, G. A. Tolstikov, and L. M. Zelenova, Izv. Akad. Nauk SSSR, Ser. Khim., 2018 (1975).
- 5. E. Klein, F. Thömel, M. Struwe, P. Heimbach, and H. Schenkluhn, Ann. Chem., 352 (1976).
- 6. C. Descoins and C. A. Henrick, Tetrahedron Lett., No. 3, 2999 (1972).
- 7. G. A. Tolstikov, U. M. Dzhemilev, and R. I. Khusnutdinov, Zh. Obshch. Khim., 45, 1322 (1975).
- 8. M. Tamura and J. Kochi, J. Am. Chem. Soc., 93, 1485 (1971).